

Nuclear Magnetic Moments

Any motion of a charged has an associated magnetic field. This on a combined scale an electrical current which is due to motion of e^- along the conductor, produces such a field.

- Electrons or nucleus possess angular momentum there is a magnetic moment.

Suppose an electron is travelling in an orbital at an angular velocity ω .

$$i = \frac{e\omega}{2\pi}$$

$e \rightarrow$ magnitude of charge in electron

$$\text{or } i = \frac{e\omega}{2\pi} = \frac{eP}{2\pi m_e r^2}$$

$P \rightarrow$ orbital angular momentum

$m_e \rightarrow$ mass of electron

$r \rightarrow$ distance from nucleus

The magnetic moment μ generated by such motion is given in electromagnetic theory

$$\mu = \cancel{A} \cdot \cancel{A} \quad \mu = A \cdot i$$

$A \rightarrow$ Area marked out by the orbital

if e^- moves in a circular orbit

$$A = \pi r^2$$

$$\mu = -\left(\frac{e}{2m_e}\right) P$$

As per the above equation the angular momentum

is quantized in unit of \hbar . Electronic magnetic moments are quantized in units of $\frac{e\hbar}{2m_e}$ ($\hbar = \frac{h}{2\pi}$) = μ_B

$$\mu_B \rightarrow \text{Bohr's magneton} = 9.27410 \times 10^{-24} \text{ JT}^{-1}$$

Due to spin motion involved we introduced magnetic moment and angular momentum.

Hence we introduce a new factor g (Landé's splitting factor)

Thus we rewrite

$$\mu = -g \left(\frac{e}{2m_e} \right) P = -g \mu_B \frac{P}{\hbar}$$

↳ Coupled spin-orbital motion.

The constant of proportionality relating μ & P depends on the mass of the particle and charge.

$$\mu \propto \left(\frac{ze}{2m_N} \right) P$$

$m_N \rightarrow$ mass of nucleus
and $z \rightarrow$ charge number

$$\mu_N = \frac{e\hbar}{2m_p}$$

$m_p \rightarrow$ mass of proton

$$\mu_N = 5.05095 \times 10^{-27} \text{ JT}^{-1}$$

$$\mu_N = \frac{g_N \mu_N P}{\hbar}$$

$g_N \rightarrow$ nuclear g -factor
↳ $g \rightarrow$ positive for electron

Negative nuclear g -factors imply that spin magnetic moment is antiparallel to the angular momentum positive values indicate that μ & P are parallel.

- Nuclear magnetic moments are expressed as magnetogyric ratio γ .

$$\gamma = \frac{\text{magnetic moment}}{\text{angular momentum}} = \frac{\mu}{P}$$

$$\text{or } \mu = \gamma P$$

$$\gamma \hbar = g_N \mu_N$$

$$\mu = g_N \mu_N [I(I+1)]^{1/2}$$

$$\boxed{\mu = \gamma \hbar [I(I+1)]^{1/2}}$$